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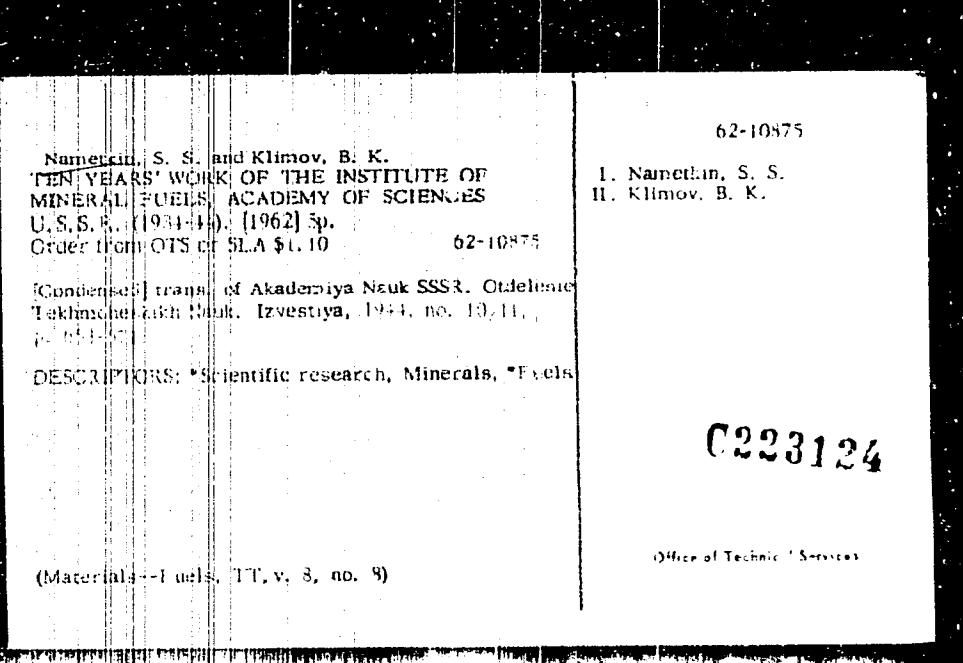
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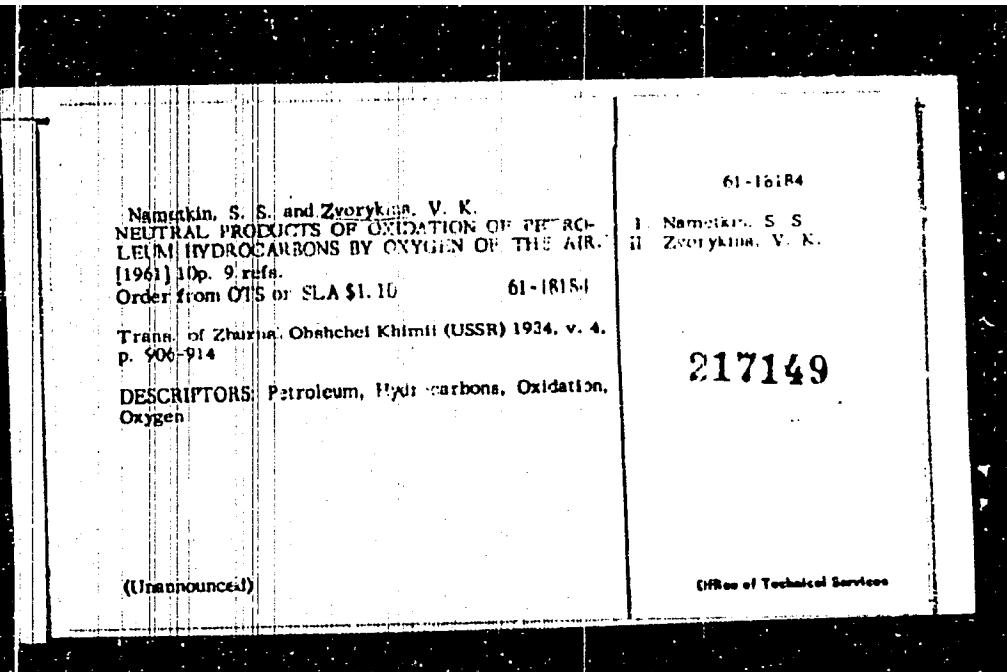
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HYDROPOLYMERIZATION OF 3-METHYL-1-BUTENE BY 100 PER CENT SULFURIC ACID. [1961]
4p. 5 refs.

Order from OTS or SLA \$1.10 61-20129

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v. 15, p. 358-361.

DESCRIPTORS: *Sulfuric acid, *Butenes, Polymers,
*Polymerization, *Hydrogenation, Hydrocarbons,
Ethylenes, Methyl radicals.

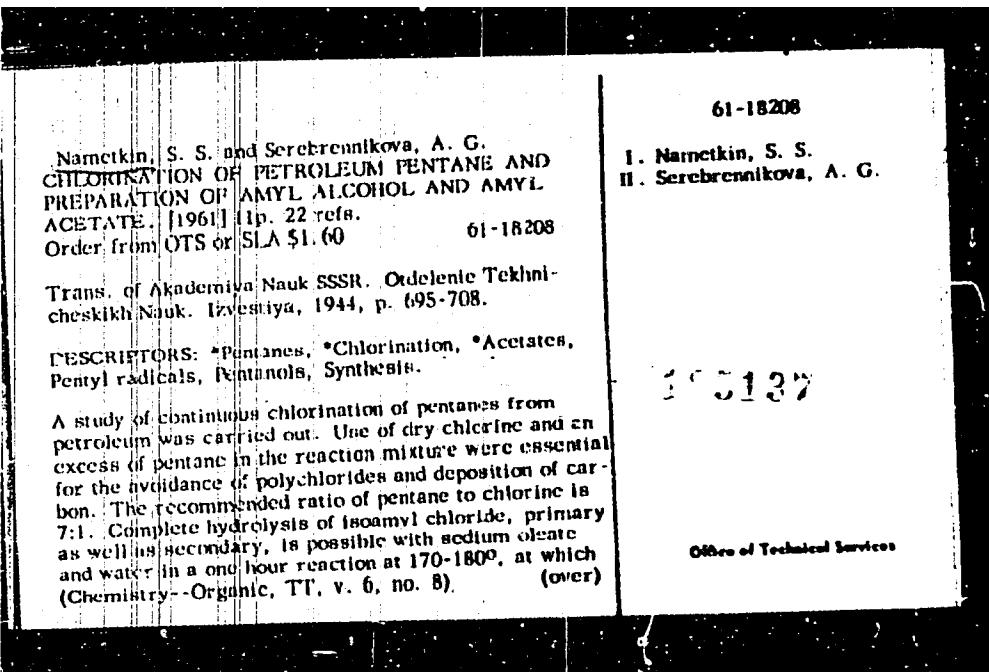
Hydropolymerization of 3-methyl-1-butene under the
action of 100% sulfuric acid was studied under different
conditions. Formation in this reaction of hydrogenated
polymers from dimers to hexamers, inclusive,
was shown. When 100% sulfuric acid is used, the hy-
dropolymerization of 3-methyl-1-butene forms more
(Chemistry--Organic, TT, v. 7, no. 5) (over)

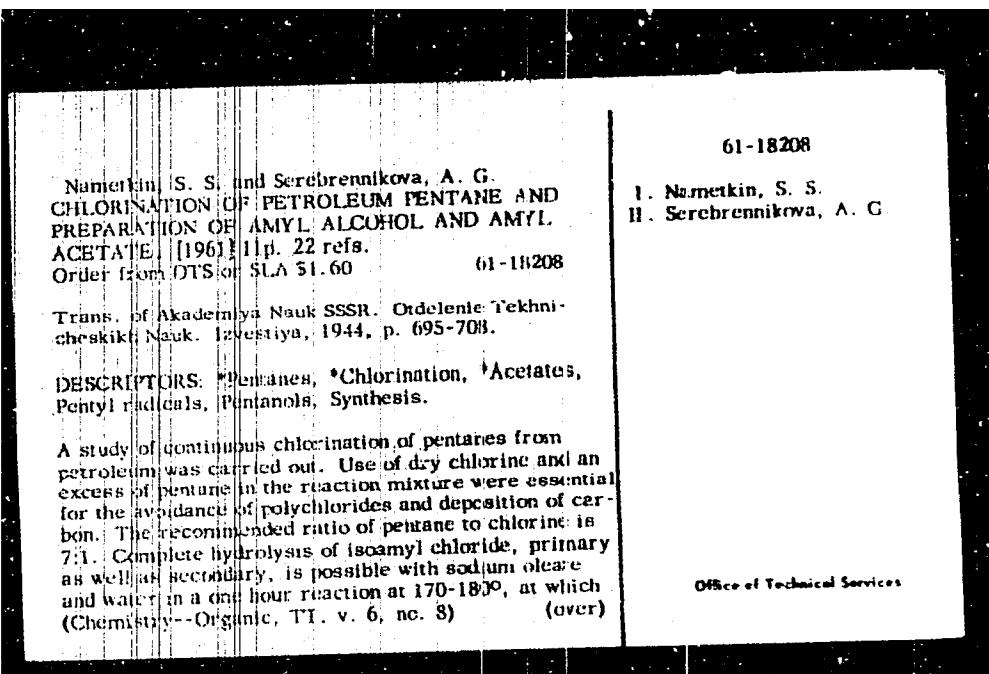
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PRELIMINARY REPT. [1963] 10p. (formulas omitted)
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Order from OTS or SLA \$1.10 63-14991

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DESCRIPTORS: "Indene," Nitric acid, "Nitration,
Hydrogenation, Chemical reactions, Chemical
properties.

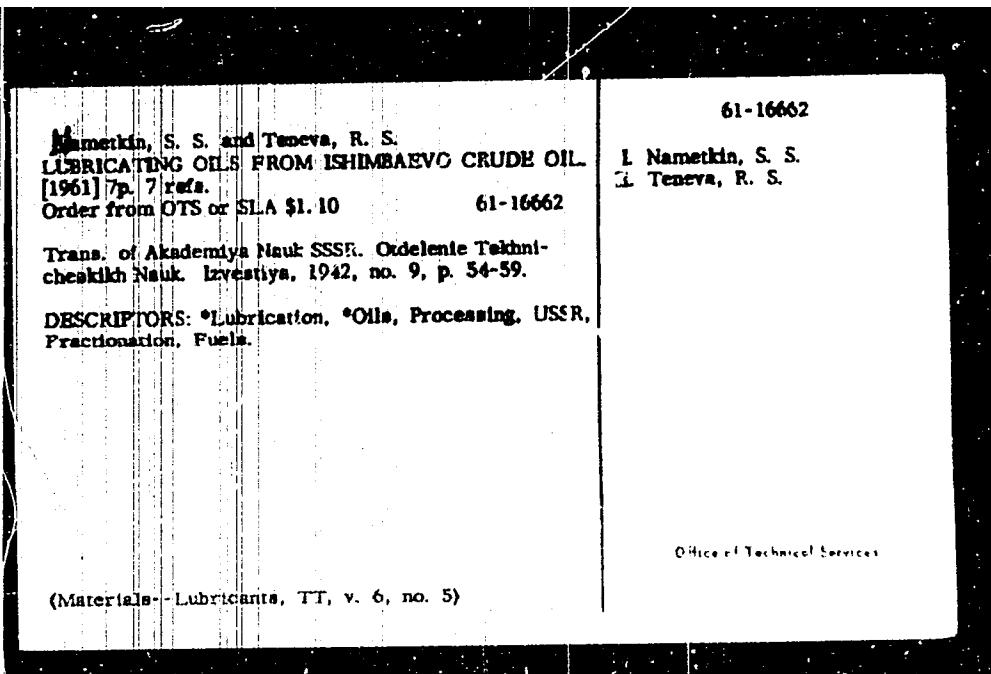
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Trans. of Akademiya Nauk SSSR. Otdelenie
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p. 10-22.

DESCRIPTORS: *Petroleum, *Gasoline, *Sulfur
compounds, Chemical analysis, Thiols, Cadmium
compounds, Chlorides, Thiophenes, Sodium compounds,
Carbamates.

A criticism (sic) of the existing methods of analysis of
sulfur compounds in gasoline established that the
methods of Faragher and Brown req. re further de-
velopment to make them more precise. Cadmium
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61-16945

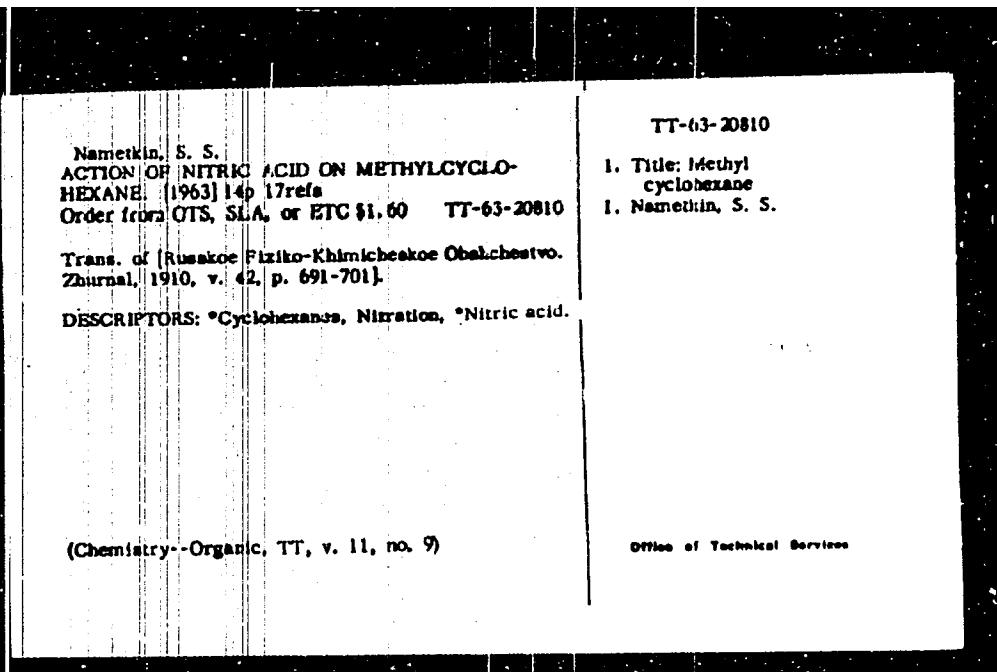
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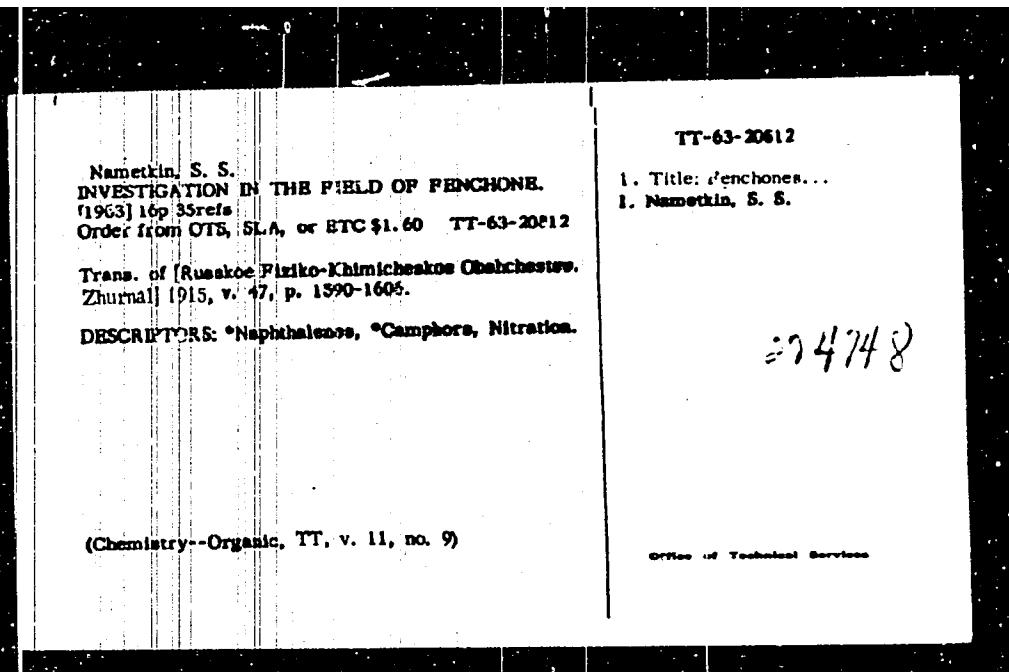
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DESCRIPTORS: *Hydrocarbons, Decomposition, Oxidation.

The basic reaction in oxidizing cracking of tetralin in the vapor phase is dehydrogenation without destruction of the ring; it results in formation of naphthalene and water. Four reactions occur simultaneously: (1) rupture of the naphthenic ring, forming aromatic hydrocarbons (2) oxidation of tetralin to the peroxide with subsequent decomposition of the latter, forming α -naphthol; (3) oxidation of toluene, forming phenoxy, and (4) (Chemistry--Organic, IT, v. 6, no. 6) (over)

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PHASE, I. [1961] 14 p. 11 refs.

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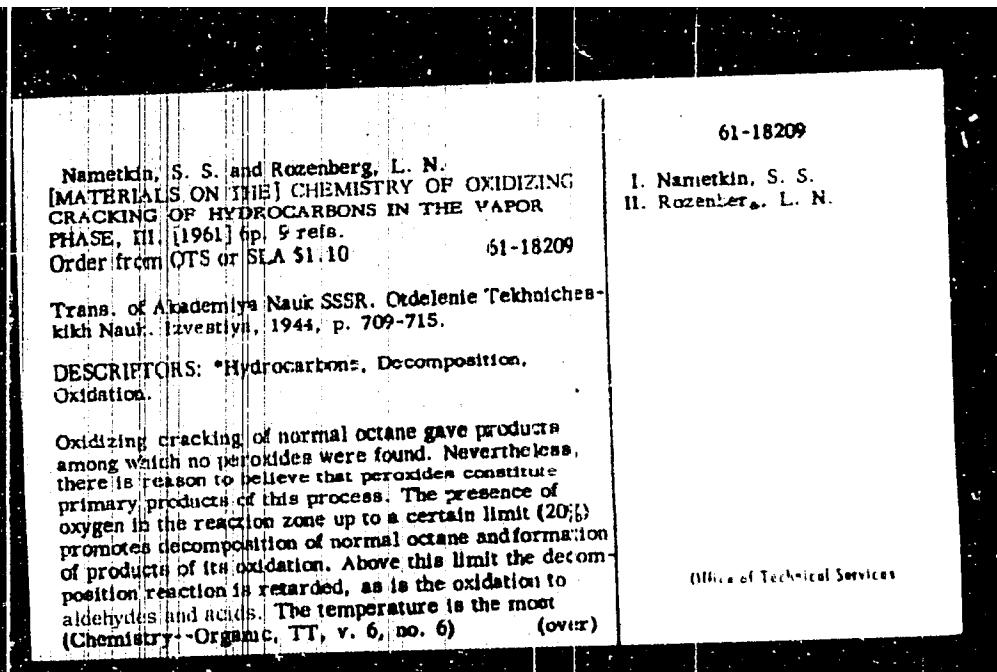
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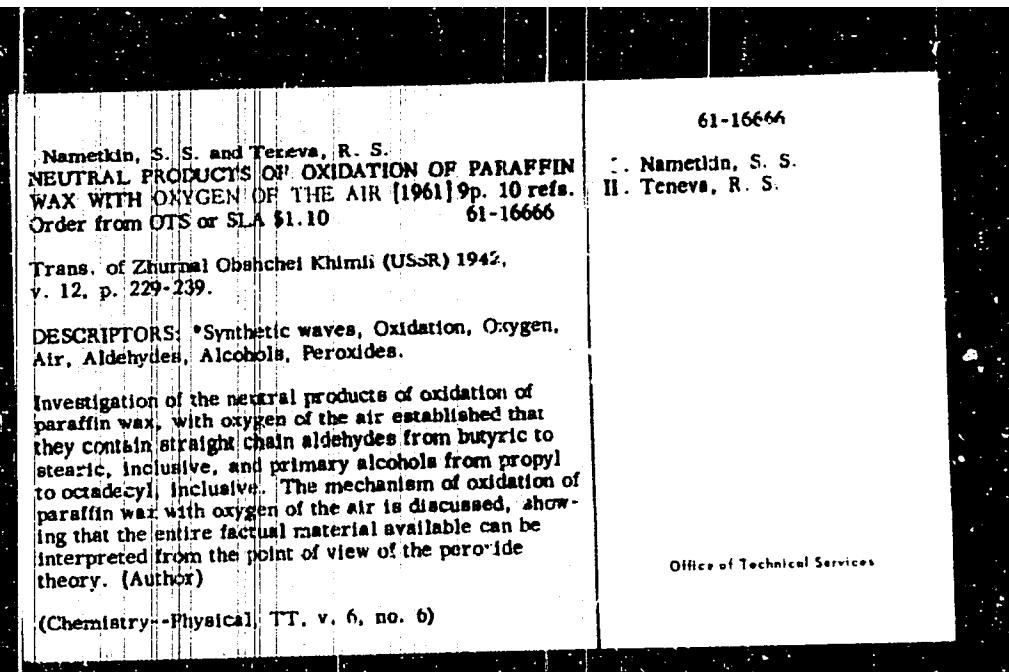
DESCRIPTORS: "Hydrocarbons, Decomposition, Oxidation."

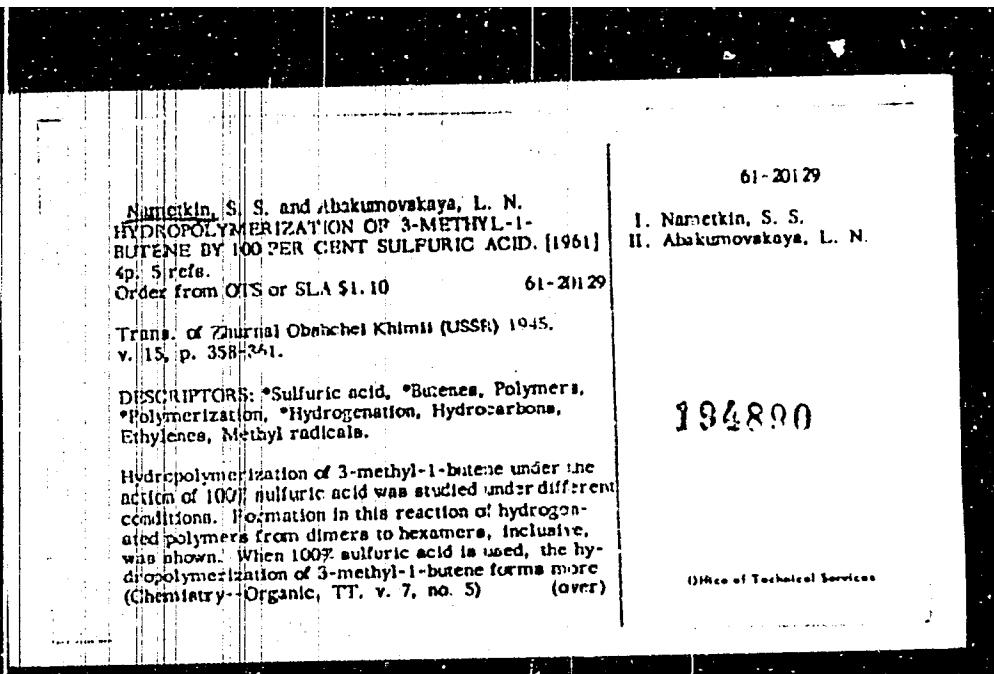
The presence of oxygen in the zone of cracking of toluene leads to oxidation of the methyl group without affecting the benzene ring. Oxidation of the side chain is apparently due to the existence of an active tautomeric form of toluene. Oxidation of toluene proceeds through formation of an intermediate peroxide compound with subsequent decomposition of the latter into benzaldehyde and water. Under conditions of oxidizing cracking in the vapor phase, toluene yields diphenyl- (Chemistry--Organic, TT, v. 6, no. 6) (over)

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Sex: M
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Steel, Machines, Design, Construction, Castings,
Steel industry, Industrial plants.

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